

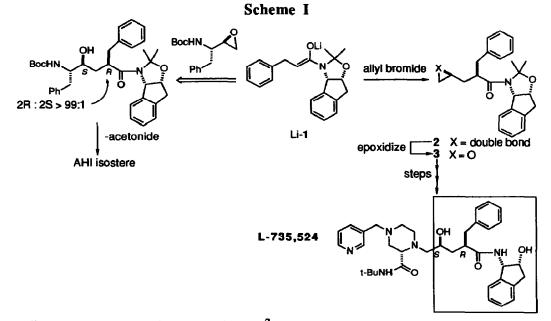
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Highly Diastereoselective Reaction of a Chiral, Non-Racemic Amide Enolate with (S)-Glycidyl Tosylate. Synthesis of the Orally Active HIV-1 Protease Inhibitor L-735,524

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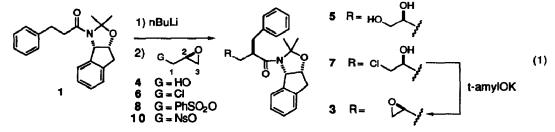
Summary: Reaction of chiral amide enolate Li-1 with (S)-glycidyl tosylate 11 affords the epoxide 3 in 72% yield with high diastereoselectivity. Epoxide 3 is converted to the orally-active HIV-1 protease inhibitor L-735,524 in 71% isolated yield.

Hydroxyethylene dipeptide isosteres which contain the 1S,2R-1-amino-2-hydroxy-indanamide (AHI) moiety can be potent inhibitors of HIV-1 protease.¹ A highly diastereoselective route to the 2(R)-aryl-4(S)-hydroxy AHI isosteres was recently developed via alkylation of (N-Boc)- α -amino-epoxides with the n-BuLi generated enolate Li-1 (Scheme I).²



The orally active HIV protease inhibitor L-735,524³ comprises the same 2(R)-aryl-4(S)-hydroxy AHI array (see box) as found in earlier inhibitors. Synthesis of epoxide 3, a key intermediate for L-735,524, was initially carried out by diastereoselective allylation of 1 to afford 2 followed by epoxidation.⁴ We now report a practical, one-step transformation of 1 to 3 which circumvents the more general problem of diastereoselective epoxidation of γ , δ -unsaturated amides with iodine.⁵

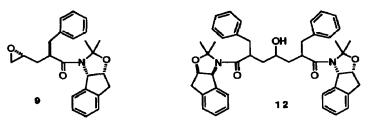
Attempted condensation of 1 with commercially available (R)-glycidol 4 via the *in-situ* enolate generation method (2 equiv. of nBuLi added to solution of 1 and 4)² failed to give any isolable coupled product 5 (eq 1). This lack of reactivity was surprising in view of the enhanced electrophilicity of alkoxy-epoxides towards nucleophilic attack.⁶ Condensation of Li-1 with (S)-epichlorohydrin 6 at -30°C gave chlorohydrin 7 as the major product in addition to epoxide 3 (7:3 \approx 2.5:1). Treatment of the mixture with base gave a 70% overall yield of 3.



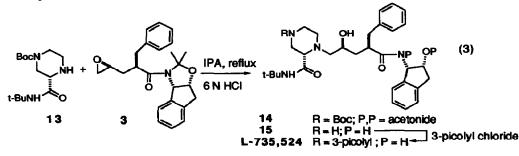
We next explored the condensation of 1 with activated glycidol derivatives as a direct route to $3.^7$ Alkylation of enolate Li-1 with the oily benzenesulfonate 8^{7c} directly afforded 3 as the major product. Epoxide 3 is derived from attack of Li-1 at the C₃ (oxirane) terminus of 8. The intermediate alkoxy-benzenesulfonate could not be detected during the course of the reaction. Also present in the crude reaction mixture was the epimeric epoxide 9 (3:9 \approx 7-8:1 by ¹H NMR analysis). Epoxide 9 is derived from *ent*-8 in the starting material, and any lack of regiochemical specificity in nucleophilic attack on 8. To improve the diastereomeric purity of the product and eliminate an extra step necessary to prepare 8, we desired a commercially available reagent of very high optical purity.

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Attempted alkylation of Li-1 with 99% ee (S)-glycidyl nosylate 10 gave a dark orange colored reaction mixture and recovered starting materials, indicative of electron transfer complications. However, condensation of amide 1 with (S)-(+)-glycidyl tosylate 11^8 of 99 % ee (1.2 equiv.) in the presence of lithium hexamethyldisilazide (LHS) gave a 72% yield of the desired epoxide 3 (eq 2). There was also produced minor amounts of the epimeric 9^9 derived from enolate attack on the C₁ terminus of 11. Thus, in a single step, two new stereocenters are introduced in extremely high diastereoselectivity. The use of the less reactive base LHS allows the coupling to be carried out by direct addition of base to a solution of 1 and 11 at -50 followed by warming to -25°C. Under these conditions, approximately 19 % of the dimeric product 12 is obtained. Crystallization of the crude mixture results in a 85% recovery of epoxide 3 in 99 % diastereomeric purity¹⁰. Not surprisingly, alkylation of 1 with (R)-glycidyl tosylate afforded the epoxide 9 as the major component.



To complete the synthesis of L-735,524, epoxide 3 and piperazine 13^{11} are subjected to a one-pot coupling/deprotection to afford 15 (eq 3). Alkylation of 15 with 3-picolyl chloride then affords L-735,524 in 71% isolated yield from 3.



In summary, we have achieved an efficient coupling of chiral amide enolate Li-1 and (S)-glycidyl tosylate 11 to afford the epoxide 3 in a single step with extremely high diastereoselectivity. This method provides a practical route to the orally active HIV protease inhibitor, L-735,524.¹²

References and Notes

(1) Lyle, T.A.; Wiscount, C.M.; Guare, J.P.; Thompson, W.J.; Anderson, P.S.; Darke, P.L.; Zugay, J.A.; Emini, E.A.; Schleif, W.A.; Quintero, J.C.; Dixon, R.A., Sigal, I.S.; Huff, J.R. J. Med. Chem. 1991, 34, 1228-1230.

(2) Askin, D.; Wallace, M.A.; Vacca, J.P.; Reamer, R.A.; Volante, R.P.; Shinkai, I. J. Org. Chem. 1992, 57, 2771-2773. The amide 1 is available from 1S,2R-1-amino-2-hydroxy-indane in an improved one-pot procedure in 90% isolated (crystallized) yield via treatment with hydrocinnamoyl chloride and triethylamine (THF) followed by 2-methoxypropene/MsOH.

(3) (a) Vacca, J.; Dorsey, B.; Levin, R.; McDaniel, S.; Darke, P.; Zugay, J.; Schleif, W.A.; Quintero, J.; Sardana, V.; Lin, J.; Chen, I-W.; Ostovic, D.; Anderson, P.S.; Emini, E.A.; Huff, J.R. *Proceedings of the National Academy of Sciences*, submitted. (b) Poster presentations at the IX International Conference on AIDS; Berlin, Germany, June 6-11, 1993 and the 206th ACS National meeting, MEDI-6, Chicago, Illinois, August 22-27, 1993.

(4) Low yields (25-35%) of epoxide 3 were obtained via per-acid treatment of olefin 2 due to non-selective epoxidation (B. Dorsey, unpublished results). A three-step epoxidation method has also been developed via asymmetric dihydroxylation of olefin 2 followed by activation and cyclodehydration (B. Dorsey, J. Vacca, to be submitted to J. Med. Chem.).

(5) Attempted diastereoselective epoxidation of 2 via iodination/iminolactone formation leads to lactonization and amide bond cleavage, cf: Tamaru, Y.; Mizutani, M.; Furukawa, Y.; Kawamura, S.; Yoshida, Z.; Yanagi, K.; Minobe, M. J. Am. Chem. Soc. 1984, 106, 1079-1085.

(6) Marshall, J.A.; Andrews, R.C. J. Org. Chem. 1985, 50, 1602-1606.

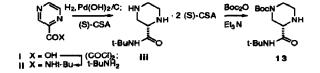
(7) (a) Klunder, J. M.; Onami, T.; Sharpless, K.B. J. Org. Chem. 1989, 54, 1295-1304. (b) Hanson, R.M. Chem. Rev. 1991, 91, 437-475. (c) Derived (PhSO₂Cl, Et₃N/CH₂Cl₂) from 83% ee 4. The benzenesulfonate was initially chosen due to the acidity of the toluenesulfonyl-methyl group. As an approximation, the pKa of p-PhSO₂C₆H₄CH₃ is 30 as measured in DMSO solution: Bordwell, F.G. Acc. Chem. Res. 1988, 21, 456-463.

(8) Available in bulk quantities in 95 % ee from D.S.M. Andeno (Netherlands) or 99% ee from DAISO (Japan). For a recent example of boron trifluoride assisted condensation of (S)-glycidyl tosylate with a lithiated sulfone, see: Baldwin, J.E.; Adlington, R.M.; Bebbington, D.; Russell, A.T. J. Chem. Soc., Chem. Commun. 1992, 1249-1251.

(9) The ratio of epoxides in the crude mixture derived from 99 % ee (S)-11 was 98:2 (3:9) as measured by normal phase HPLC: 4.6 mm x 25 cm dupont Zorbax-NH₂ column, hexane/2-propanol (98/2) mobile phase, 3.8 mL/min flow rate, 220 nm detection, approximate retention times (min): 1 = 2.0, 3 = 2.3, 9 = 3.2. Interestingly, a similar level of regiochemical control of C₃ (oxirane):C₁ attack was noted for the condensation of malonate with non-racemic epichlorohydrin: Pirrung, M.C.; Dunlap, S.E.; Trinks, U.P. Helv. Chim. Acta 1989, 72, 1301-1310.

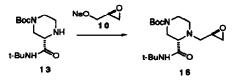
(10) Typical experimental procedure: A solution of amide 1 (1000 g, 3.11 mol) and 98 % ee 2(S)-glycidyl tosylate 11 (853 g, 3.74 mol, 1.2 equiv.) in 15.6 L of THF (KF = 22 mg/L) was degassed with nitrogen and cooled to -56°C. LHS in THF (2.6 L, 1.38 M, 1.15 equiv.) was added over 2 h, while keeping the temperature between -50 to -45°C. The reaction mixture was stirred at -45 to -40°C for 1 h and then allowed to warm to -25°C over 1 h and aged between -25 to -22°C for 4 h. The reaction was quenched with 6.7 L of water at -15°C, extracted with ethyl acetate (10 L) and washed with a solution of 1% aqueous NaHCO3 (5 L) and saturated NaCl (0.5 L). Nominal assay yield for 3 is 72% by quantitative HPLC analysis: (25 cm C₁₈ Machery-Nagel "nucleosil" column, pH 6.8 phosphate buffer/CH₃OH, 35/65 (v/v), 0-20 min, isocratic, then gradient over 20 min to 10/90 (v/v), 40°C, 220 nm detection, retention times: 3 = 32.4 min., 9 = 27.7 min. The ethyl acetate extract was concentrated *in vacuo* and solvent switched to MeOH and further concentrated to a final volume of 3.2 L and cooled to 5°C. The white slurry was filtered and the wet cake was washed with cold methanol (2 x 250 mL). The washed cake was dried under vacuum (26 in. Hg) at 25°C to afford 727 g of epoxide 3 (61 % isolated yield, 98.7 area % 3, 1.3 area % 9 by HPLC analysis). Spectral data for 3: ¹³C NMR (300 MHz, CDCl3) δ 171.1, 140.6, 140.5, 139.6, 129.6, 128.8, 128.2, 127.2, 126.8, 125.6, 124.1, 96.8, 79.1, 65.5, 50.5, 47.7, 45.9, 37.8, 37.1, 36.2, 26.6, 23.8.

(11) (a) Piperazine 13 was prepared in 26% overall yield from 2-pyrazinecarboxylic acid by the 4-step procedure:



(b) Felder, von E.; Maffei, S.; Pietra, S.; Pitrè, D. Helv. Chim. Acta 1960, 43, 888-896.

(12) An alternate route to L-735,524 would involve initial connection of the non-racemic glycidyl unit with the piperazine 13. Attempted coupling of 13 with tosylate 11 afforded a complex reaction mixture in which the major product did not contain an epoxide moiety. However, coupling of 13 with the more C₁-reactive (S)-glycidyl nosylate 10 afforded the desired piperazine-epoxide 16 in 72% isolated yield. Unfortunately, the coupling of Li-1 and 16 afforded intermediate 14 in only 20% yield.



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